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Supplementary Materials



Fig. S1. Schematic diagram of the experimental setup of E-Fenton.



Fig. S2. Normal plots of residuals (a, d), Residuals vs. predicted plots (b, e), Predicted vs. actual plots (c, f) for % COD degradation and energy consumption, respectively.



Fig. S3. Proposed pathway for the degradation of electroplating effluents by E-Fenton process.



Fig. S4. (a) The average degradation of COD by GO/TiO₂NTs electrodes in the presence of scavengers and (b) Degradation of electroplating effluents in the presence of different scavengers under optimal operational parameters.



Fig. S5. Reusability and stability of GO/TiO2NTs electrode up to thirty consecutive cycles.

Factor	Parameters	Range of actual and coded variables			
ractor		Coded Low level (-1)	Coded (0)	Coded High level (+1)	
А	Time; t (min)	30	90	150	
В	Current; i (Ampere)	0.4	1.0	1.6	
С	Ferrous Sulphate Concentration; (CFe) (mM)	0.2	0.6	1.0	

Table S1. Experimental design levels of chosen parameters

Table S2. ANOVA suggested by BBD for the % COD removal and energy consumption

	% COD Degradation					Energy Consumption				
Source	SS	DF	MS	F-value	p-value	SS	DF	MS	F-value	p-value
Model	SS	DF	MS	F-value	p-value	19.77	9	2.20	273.73	< 0.0001
A-Time	299.50	9	33.28	32.75	< 0.0001	5.19	1	5.19	646.60	< 0.0001
B-Current	62.83	1	62.83	61.84	0.0001	13.60	1	13.60	1694.11	< 0.0001
$C-FeSO_4$	195.92	1	195.92	192.82	< 0.0001	0.0003	1	0.0003	0.0431	0.8415
AB	11.59	1	11.59	11.41	0.0118	0.8741	1	0.8741	108.91	< 0.0001
AC	7.51	1	7.51	7.39	0.0298	0.0040	1	0.0040	0.4943	0.5047
BC	0.4356	1	0.4356	0.4287	0.5335	0.0002	1	0.0002	0.0307	0.8658
A ²	0.0002	1	0.0002	0.0002	0.9885	0.0970	1	0.0970	12.09	0.0103
B^2	14.96	1	14.96	14.72	0.0064	0.0016	1	0.0016	0.1988	0.6692
\mathbb{C}^2	2.96	1	2.96	2.91	0.1319	0.0073	1	0.0073	0.9048	0.3732
Residual	3.47	1	3.47	3.41	0.1073	0.0562	7	0.0080		
Lack of Fit	7.11	7	1.02			0.0549	3	0.0183	56.10	0.0010
Pure Error	1.46	3	0.4873	0.3449	0.7958	0.0013	4	0.0003		
Correlation Total	5.65	4	1.41			19.83	16			

*DF- Degree of freedom; SS – Sum of squares; MS - Mean of squares.

SS	D	н	MS	F-value	p-value	Remark	SS	DF	MS	F-value	p-value	Remark
		% COD	degradation	u					Energy co	onsumption	_	
				Se	squential mo	del sum of sq	uares					
1.335E+	-05 1	1.	335E+05				81.51	7	81.51			
270.3	5 3		90.12	32.31	< 0.0001		18.79	3	6.26	78.11	< 0.0001	
7.94	ŝ		2.65	0.9350	0.4595		0.8783	3	0.2928	17.86	0.0002	
21.21			7.07	6.96	0.0166	Suggested	0.1077	3	0.0359	4.47	0.0470	Suggested
1.46	33		0.4873	0.3449	0.7958	Aliased	0.0549	3	0.0183	56.10	0.0010	Aliased
5.65	4		1.41				0.0013	4	0.0003			
1.338E+	-05 15		7869.35				101.34	17	5.96			
					Lack	of fit tests						
30.61	6		3.40	2.41	0.2062		1.04	6	0.1157	354.67	< 0.0001	
22.67	, 6		3.78	2.67	0.1801		0.1626	9	0.0271	83.10	0.0004	
1.46	33		0.4873	0.3449	0.7958	Suggested	0.0549	3	0.0183	56.10	0.0010	Suggested
0.000	0 0	-				Aliased	0.0000	0				Aliased
or 5.65	4		1.41				0.0013	4	0.0003			
					Model su	mmary statistic	Ş					
Std. d€	şν. R	7	Adj. R ²	Pre. \mathbb{R}^2	Press	Remark	Std. dev.	\mathbb{R}^2	Adj. R ²	Pre. R ²	Press	Remark
1.67	0.85	317	0.8544	0.7835	66.38		0.2831	0.9474	0.9353	0.8917	2.15	
1.68	0.90	176	0.8522	0.6546	105.90		0.1280	0.9917	0.9868	0.9646	0.7027	
c 1.01	0.97	768	0.9470	0.8949	32.22	Suggested	0.0896	0.9972	0.9935	0.9556	0.8801	Suggested
1.19	0.98	316	0.9263			Aliased	0.0181	0.99999	0.9997			Aliased

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Responses	\mathbf{R}^2	Adj R ²	Pred R ²	Adeq Precision
Z_1	0.9768	0.9470	0.8949	21.0112
Z_2	0.9972	0.9935	0.9556	61.3909

Table S4. Various R^2 values proposed by BBD for responses % COD degradation (Z₁) and energy consumption (Z₂)

 $Z_1: \ \% \ \text{COD} \ \text{Removal}; \ Z_2: \ \text{Energy} \ \text{Consumption} \ (kWh/m^3); \ R^2: \ R-squared; \ \text{Adj: Adjusted}; \ \text{Pre- Predicted}; \ \text{Adeq: Adequate.} \ Adequate.$

S.No.	Compound	Formula	Retention Time (min)	Molecular Mass	Matching %
1	(3-Methyl-oxiran-2-yl) methanol	$C_4H_8O_2$	1.653	88.11	68.08
2	Methylazoxymethanol acetate	$C_4H_8N_2O_3$	1.710	132.11	71.47
3	Propanedioic acid	$C_3H_4O_4$	2.057	104.06	76.89
4	Azodicarbonamide	$C_2H_4N_4O_2$	2.157	116.08	75.72
5	Propanoic acid, ethyl ester	$C_5H_{10}O_2$	2.994	102.13	91.97
6	Ethylene glycol acetate formate	$C_5H_8O_4$	5.120	132.11	75.73
7	1-Butanamine, Nmethyl-	$C_5H_{13}N$	9.509	87.16	69.33
8	n-Hexylmethylamine	C7H17N	15.201	115.22	75.17
9	Oxalic acid	$C_2H_2O_4$	16.531	90.03	68.62
10	1,2-Ethanediamine, N,N'-dimethyl-	$C_4H_{12}N_2$	19.111	86.13	66.67
11	1-Octanamine, Nmethyl-	C_9H_21N	20.889	143.27	70.25
12	1-Methyldodecylamine	$C_{13}H_{29}N$	22.139	199.38	71.87
13	Methylpent-4-enylamine	$C_6H_{13}N$	22.139	99.17	70.62
14	1-Dodecanol	$C_{12}H_{26}O$	22.340	186.33	89.34
15	1-Nonanol	$C_9H_{20}O$	22.632	144.25	70.08
16	1-Decanol	$C_{10}H_{22}O$	25.153	158.28	79.94
17	Phenethylamine, p,α- dimethyl-	$C_{10}H_{15}N$	25.821	149.23	81.23
18	dl-Alanine	$C_3H_7NO_2$	27.709	89.09	79.27
19	Dibutyl phthalate	$C_{16}H_{22}O_4$	27.827	278.34	87.97
20	Phthalic acid, butyl hex- 3-yl ester	$C_{18}H_{26}O_4$	29.354	306.39	85.62
21	Sarcosine ethyl ester hydrochloride	$C_5H_{11}NO_2$	30.699	153.61	78.38

Table S5. List of various compounds identified with GC - MS analysis in untreated electroplating wastewater

Table S6. List of various compounds identified with GC - MS analysis in E-Fenton treated electroplating wastewater

S.No.	Compound	Formula	Retention Time (min)	Molecular Mass	Matching %
1	D-Allothreonine	$C_4H_9NO_3$	2.199	119.12	78.61
2	1,2-Dibutoxyethane	$C_{10}H_{22}O_2$	2.301	174.28	72.18
3	Di-n-propyl ether	$C_6H_{14}O$	2.820	102.17	74.77
4	Amphetamine	$C_9H_{13}N$	3.000	135.21	85.78
5	1-Propanamine, N,2-dimethyl	$C_5H_{13}N$	9.517	87.163	75.23
6	l-Alanine ethylamide	$C_5H_{12}N_2O$	11.211	187.24	81.17
7	1-Octanamine, N-methyl-	$C_9H_{21}N$	15.643	143.26	89.39
8	2-Butanamine, 3-methyl-	$C_5H_{13}N$	20.889	87.16	88.16
9	2-Heptanamine, 5-methyl-	$C_8H_{19}N$	21.634	129.25	77.02
10	Amphetamine-3-methyl	$C_{10}H_{15}N$	25.826	149.23	77.37
11	Benzenemethanol, α -(1-aminoethyl)-	$C_9H_{13}NO$	28.443	151.20	84.43
12	Dibutyl phthalate	$C_{16}H_{22}O_4$	31.827	278.34	87.05