



Supplementary Materials

Table S1. Textural and Optic Properties of the Synthesized Catalysts

Sample	Crystallite size (nm)	Band gap (eV)	Surface area ($\text{m}^2 \text{ g}^{-1}$)	Ionic radius (Å)	EDX composition
TiO ₂	3.91	3.03	5.71	0.68	O (46.53%) Ti (1.45%)
Ni-TiO ₂	2.96	2.54	12.19	0.72	O (94.98%) Ti (3.42%) Ni (1.61%)

Table S2. Composition of Formulated Wastewater

Parameters	Value
CHD concentration (mg/L)	30
pH	6.8
TOC (mg/L)	34.56
COD (mg/L)	300
BOD ₅ (mg/L)	280
UV ₂₅₄ absorbance	1.35

Table S3. ANOVA Analysis for Catalytic Activity of Different Photocatalyst under Visible Light

Photocatalyst	Source	Sum Square	df	Mean Square	F-value	P-value	R ²	adj R ²
TiO ₂	Quadratic	901.33	9	100.15	27.84	< 0.0001		
	Residual	35.97	10	3.60	-	-		
	Lack of Fit	27.32	5	5.46	3.16	0.1164	0.9616	0.9271
	Pure error	8.65	5	1.73	-	-		
	Corr. Total	937.30	19	-	-	-		
Ni-TiO ₂	Quadratic	13,314.63	9	1,479.40	396.87	< 0.0001		
	Residual	40.00	10	-	-	-		
	Lack of Fit	21.90	5	4.38	1.21	0.4196	0.9970	0.9943
	Pure error	18.10	5	3.62	-	-		
	Corr. Total	13,354.54	19	-	-	-		

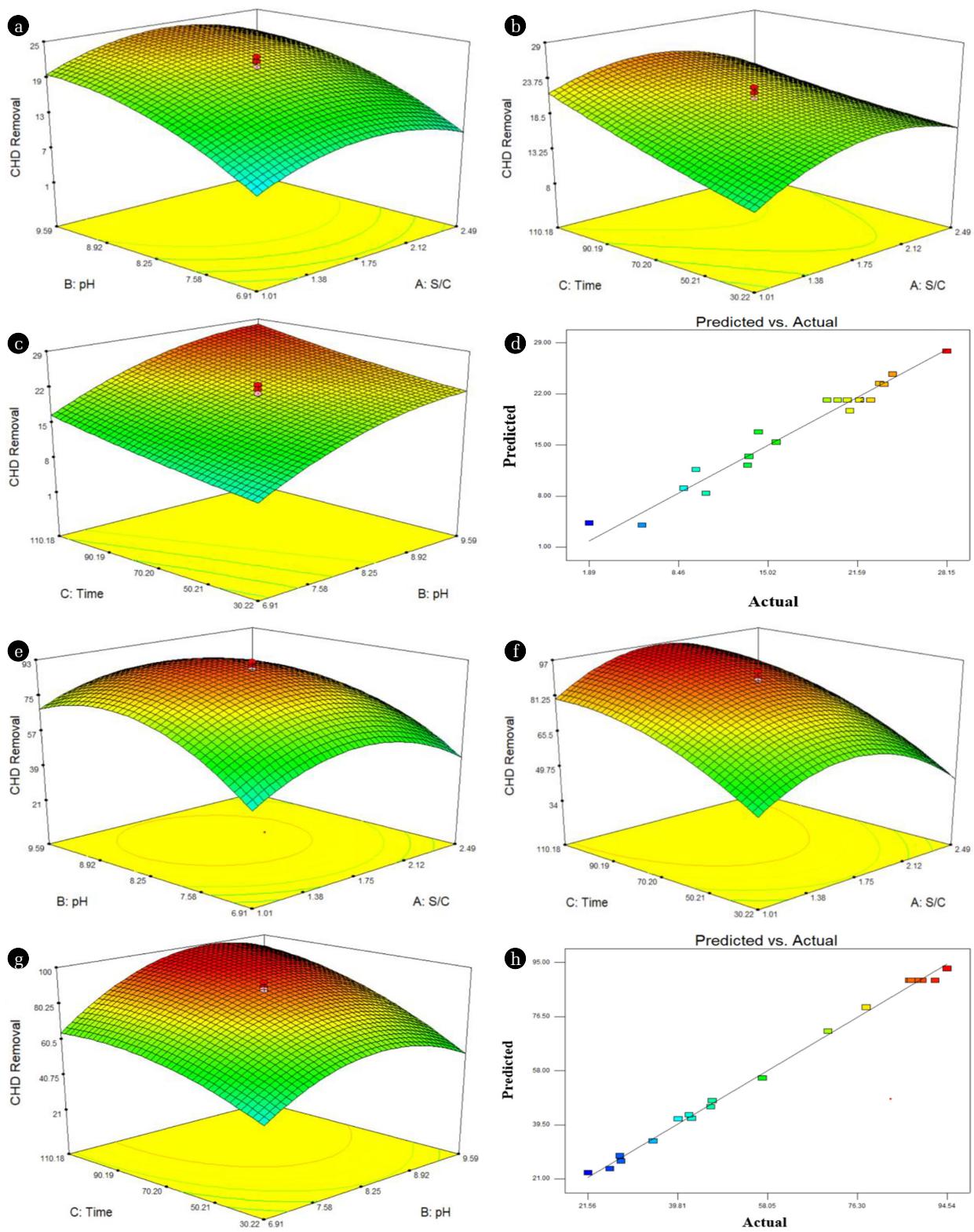


Fig. S1. Graphical representation of interaction among (a) pH and S/C; (b) Time and S/C; (c) Time and pH and; (d) Predicted versus Actual graph (TiO_2); (e) pH and S/C; (f) Time and S/C ; (g) Time and pH and ; (h) Predicted versus Actual graph (Ni-TiO_2).

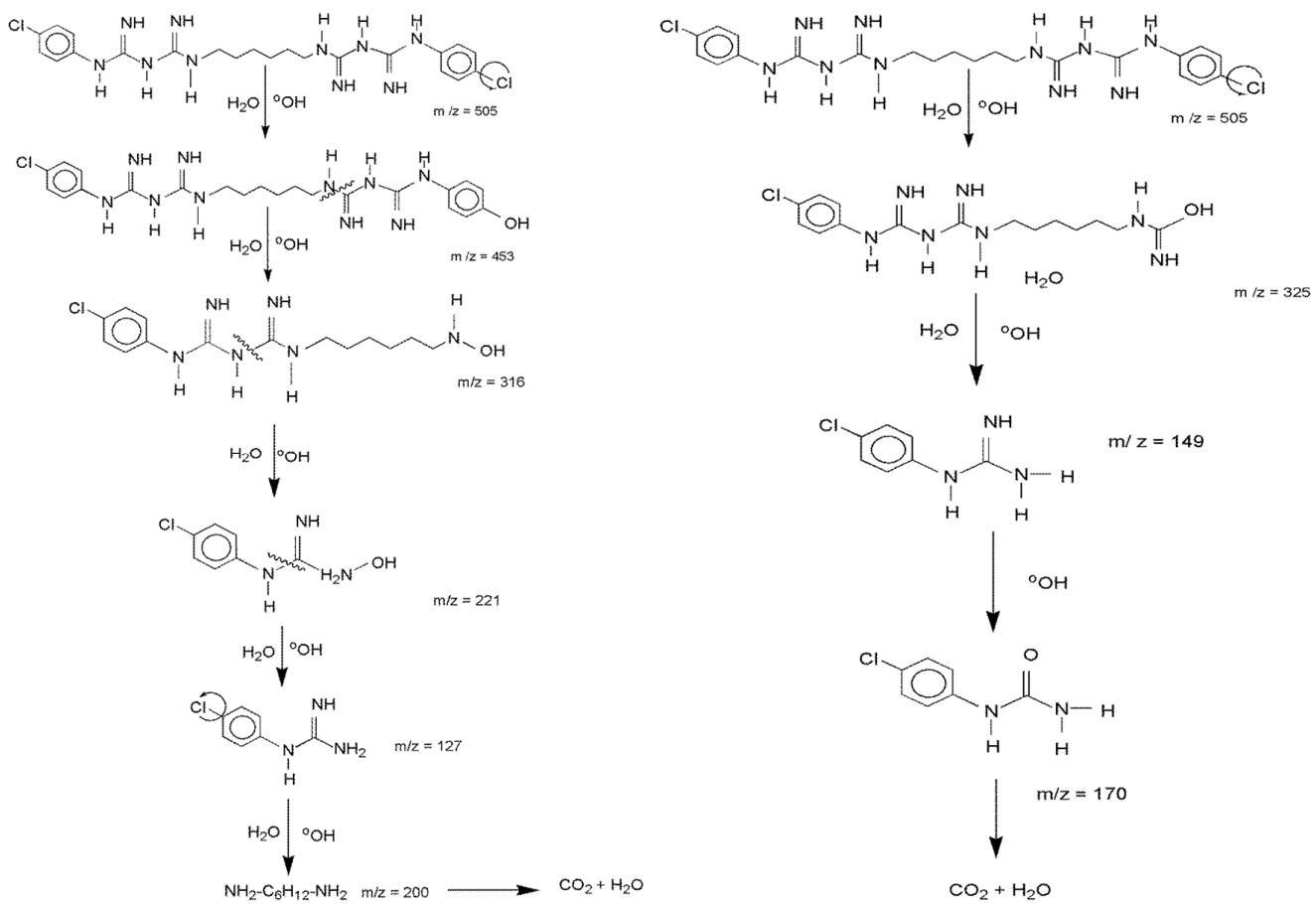
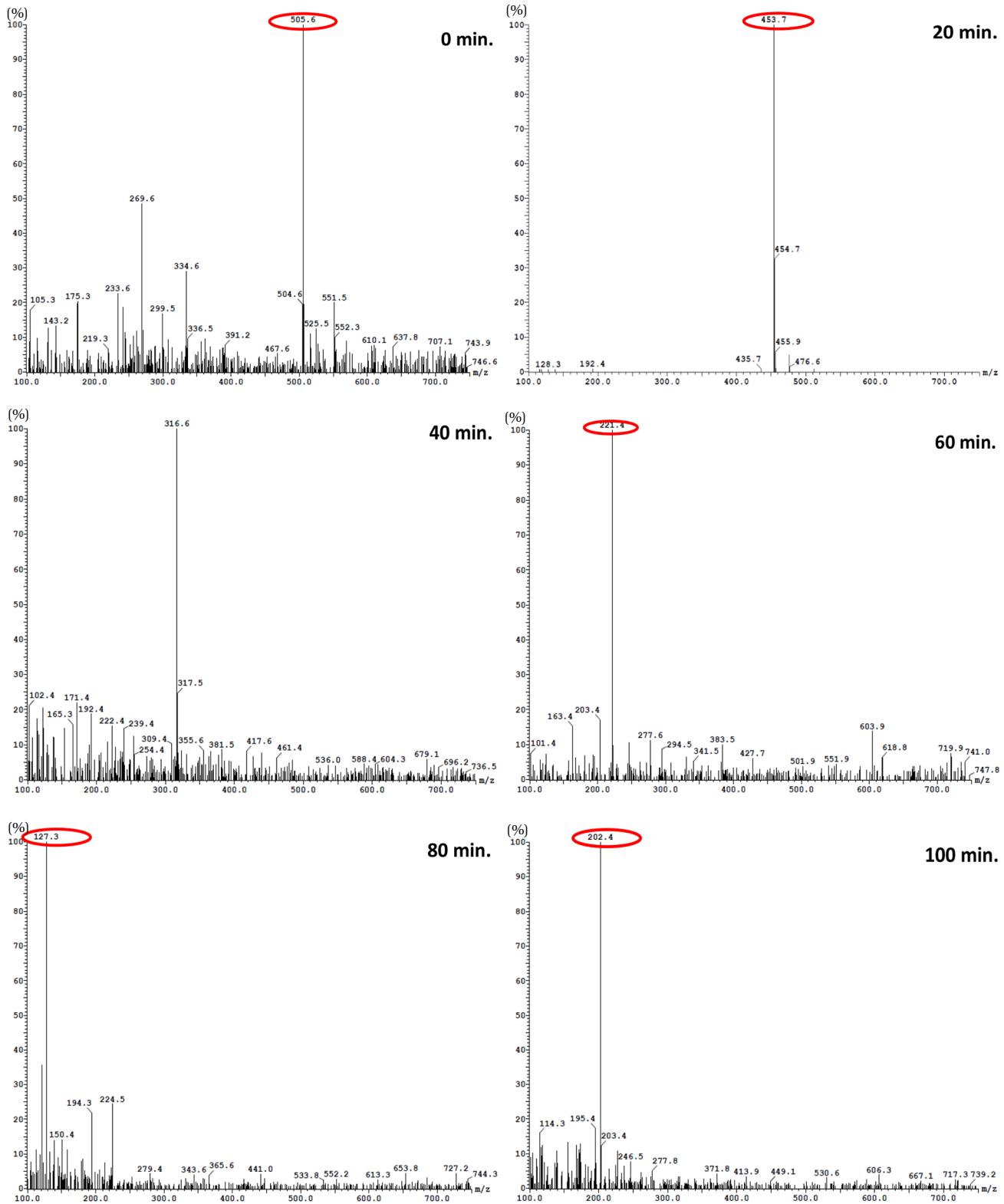


Fig. S2. Possible pathway of photocatalytic degradation of formulated wastewater using Ni-TiO₂ under (a) visible light and (b) solar light.

a



(b)

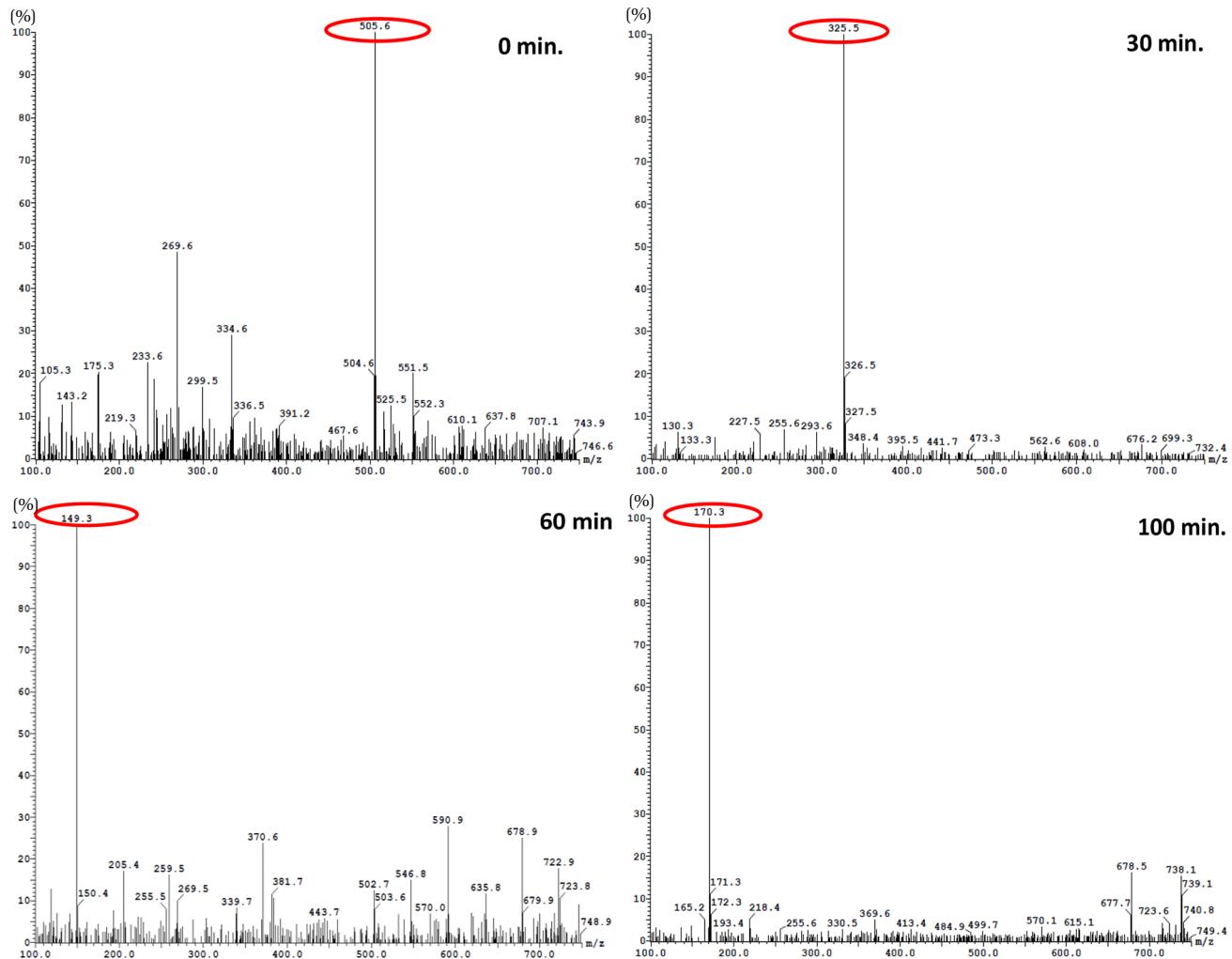


Fig. S3. LC-MS of photocatalytic degradation of formulated wastewater using Ni-TiO₂ (a) under visible light; (b) under solar light.

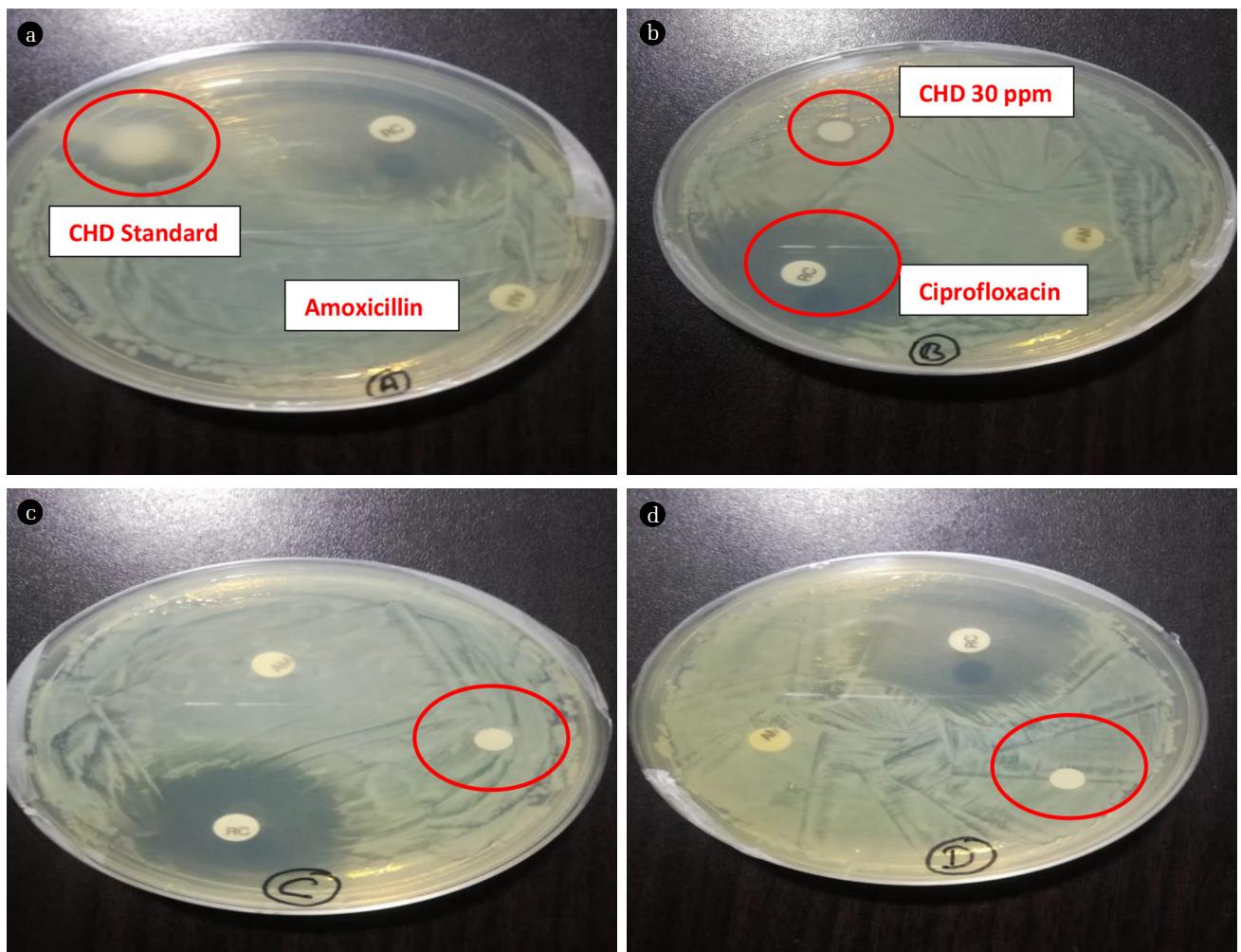


Fig. S4. Bacterial susceptibility test on bacterial strains, (a) Standard CHD; (b) CHD solution (30 ppm); (c) Formulated waste treated by Ni-TiO₂ under visible light; and (d) Formulated waste treated by Ni-TiO₂ under solar light.

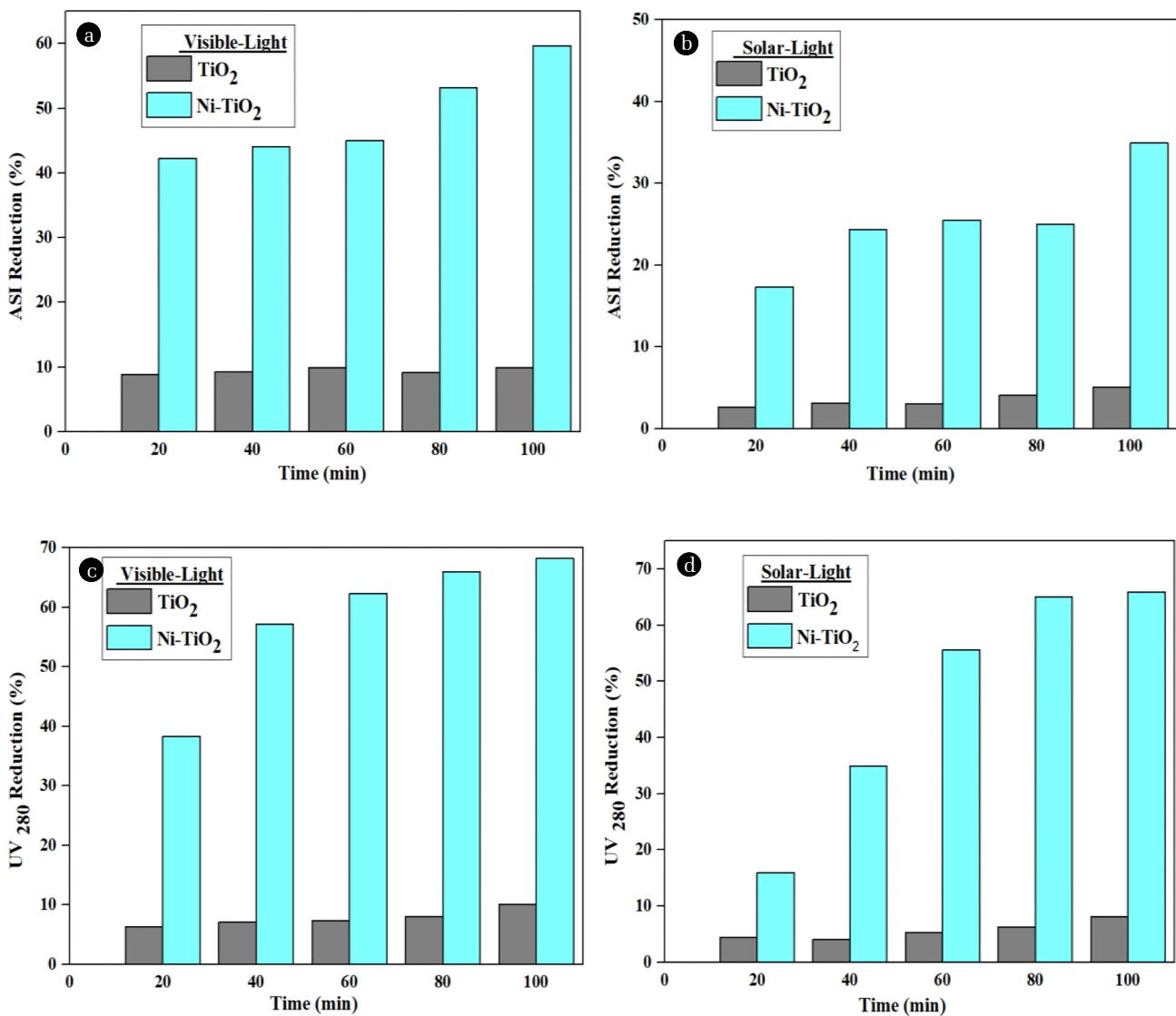
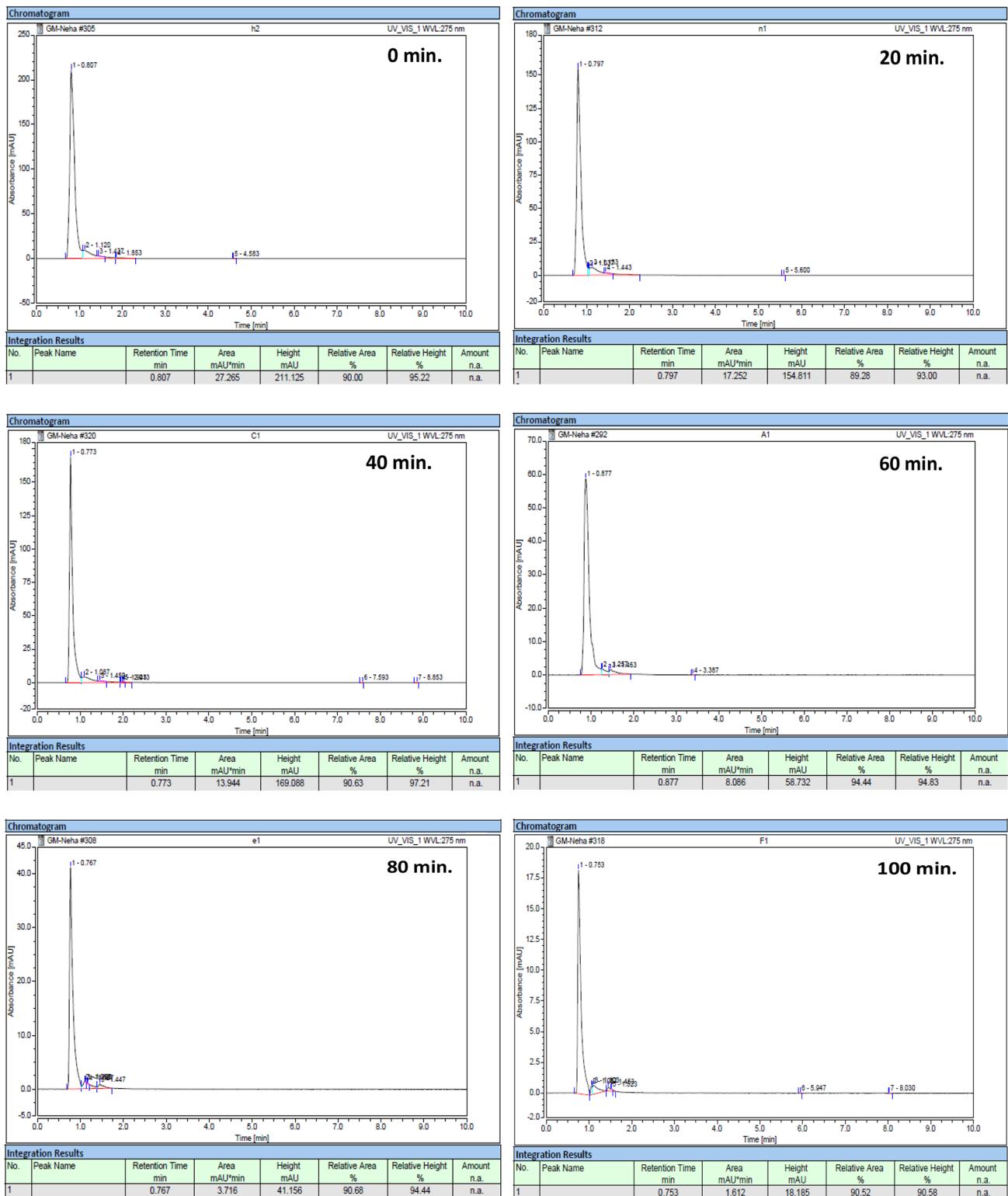


Fig. S5. Reduction in ASI values (a), (b) and UV280 (c), (d) during mineralization of CHD using TiO₂ and Ni-TiO₂ under solar and visible light.

a



b

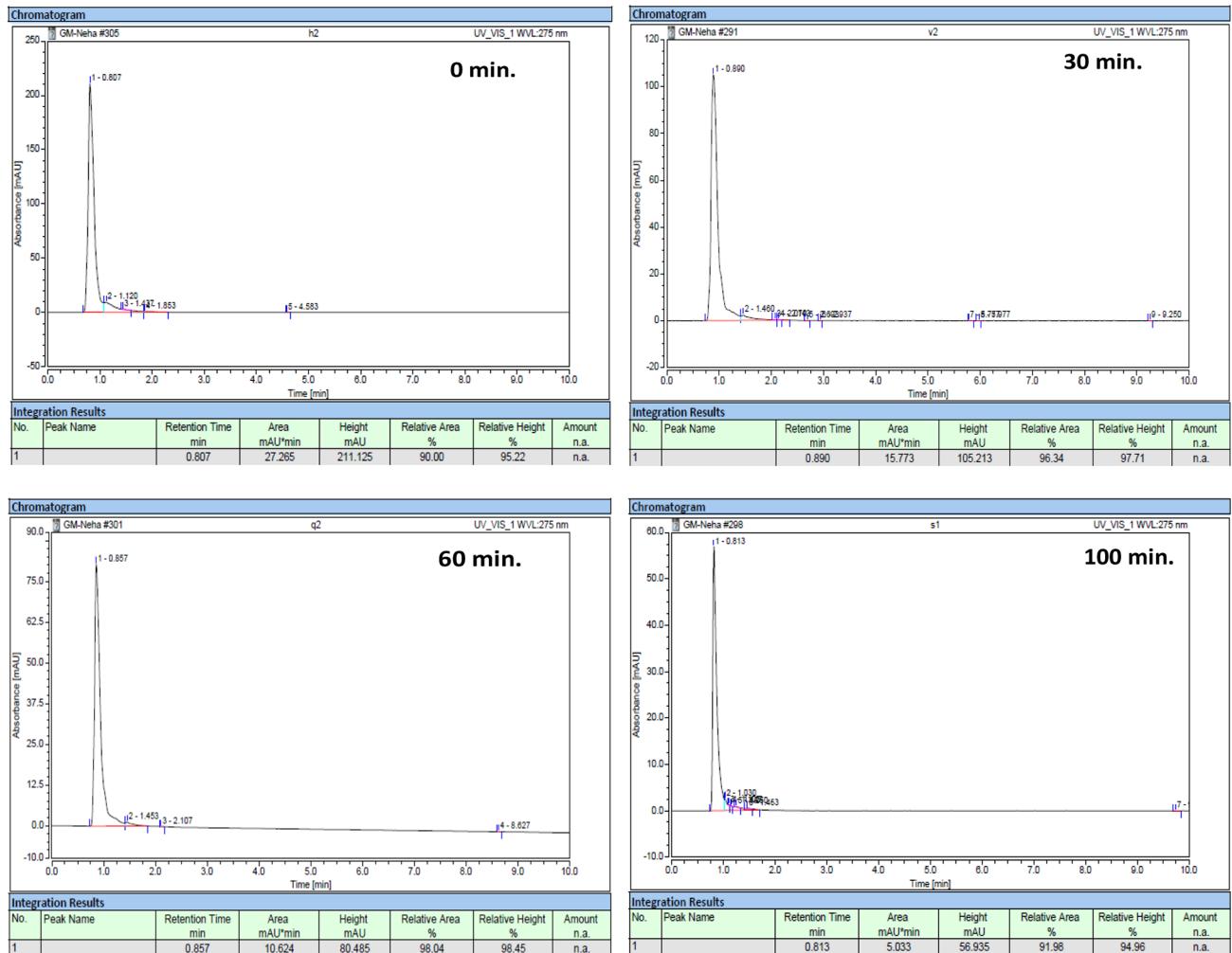


Fig. S6. HPLC of photocatalytic degradation of formulated wastewater using Ni-TiO₂ (a) under visible light; (b) under solar light.